

Homoepitaxial growth of large-area rhombohedral-stacked MoS₂

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Interlayer stacking is an important degree of freedom to tune the properties of two-dimensional materials and offers enormous opportunities for designing functional devices. As a classic example, rhombohedral-stacked (3R) two-dimensional materials exhibit ferroelectricity and optical nonlinearity that are non-existent in naturally abundant hexagonal-stacked (2H) counterparts. However, the ability to grow stacking-controlled large-area films remains challenging due to the thermodynamic competition of different polytypes. Here we report the chemical vapour deposition growth of two-inch wafer-scale 3R-MoS₂ films with high phase purity by homoepitaxy on top of a crystalline monolayer MoS₂. A defect-promoted nucleation mechanism was proposed, in which Mo-substituted sulfur vacancy is identified as one of the possible defects promoting 3R stacking. We fabricate ferroelectric semiconductor field-effect transistors with 3R-MoS₂ channels and demonstrate non-volatile memory characteristics. The control of interlayer stacking is an essential step towards the large-scale production of two-dimensional materials for multifunctional integration.

Future electronics beyond Moore's path require multifunctional integration, where materials and devices with different functionalities, such as memory, sensor and optoelectronics, are all integrated monolithically¹. Two-dimensional (2D) transition metal dichalcogenides (TMDs) are important material candidates for transistor scaling and monolithic three-dimensional integration^{2,3}. To achieve this goal, we need to engineer materials with the desired properties, and interlayer stacking is one of the most important tuning parameters for 2D materials. In fact, many exotic properties are observed in artificially stacked multilayer 2D materials that are non-existent in naturally abundant counterparts, such as ferroelectricity^{4–8}, superconductivity⁹, topological insulator¹⁰ and valleytronics^{11,12}. However, early proof-of-concept demonstrations rely on either mechanically stacked moiré structures⁵ or preselected flakes from stacking uncontrolled chemical vapour deposition (CVD) growth^{13,14}. The lack of interlayer stacking control

in CVD thus far poses substantial challenges in large-scale device integration.

Many semiconducting TMDs (such as MoS₂ and WSe₂) have two thermodynamically stable polytypes, namely, 2H and 3R. The 2H (3R) phase has a unit cell composed of bilayers (trilayers) with an interlayer twist angle of 180° (0°). In the 3R phase, each layer in the unit cell slides $\frac{\sqrt{3}}{3}a$ along the armchair direction (where a is the lattice constant), giving rise to inversion symmetry breaking. Recent works show the promise of 3R-TMDs in nonlinear optics¹⁵, ferroelectrics⁸ and bulk photovoltaics¹⁶ owing to their non-centrosymmetric structure. Currently, the main preparation method for 3R-TMDs relies on mechanical exfoliation from a bulk rhombohedral crystal^{11,17,18}, which suffers from a small flake size and low productivity. On the other hand, CVD is a more promising technique to produce large-area films^{19,20}, but has limited success so far in growing a stacking-controlled 3R phase^{21–23}.

Importantly, the lack of mechanistic insights to preferentially grow 3R stackings over other competing phases (especially the most thermodynamically stable and abundant phase is 2H) prevents the large-area growth of 3R-stacked TMDs^{24–26}.

In this work, we grow 3R-MoS₂ films with high phase purity (>99.86%) on a two-inch sapphire substrate by CVD. The growth is based on homoepitaxy, where the first single-crystalline MoS₂ layer serves as the template for the nucleation of upper layers in a Mo-rich environment. Combining extensive experiments with density functional theory (DFT) calculations, we propose a defect nucleation mechanism in which a Mo-substituted sulfur vacancy (Mo_s) is identified as the most likely nucleation centre for 3R-MoS₂. Such a hypothesis is corroborated by the existence of Mo_s within an early nucleation cluster captured by scanning transmission electron microscopy (STEM). We further confirm the presence of ferroelectricity in the 3R-MoS₂ films and fabricate ferroelectric semiconductor field-effect transistors (FeS-FETs) showing >10-year retention time and multibit storage capacity at room temperature.

Homoepitaxial growth of large-area 3R-MoS₂

The CVD growth was carried out in a custom tube furnace (Supplementary Fig. 1) by a two-step process, where a continuous single-crystal monolayer was formed in the first step on c-plane sapphire substrates with a miscut angle of 1° towards the *a* axis (*c/a* = 1°; ref. 27) and the upper layers were grown in the second step (Methods). During the second step, we increased the Mo concentration in situ by elevating the source temperature without breaking the vacuum. Control experiments that expose the first MoS₂ layer to ambient conditions or oxygen after the first step result in mixed 3R and 2H phases, which will be discussed later. Figure 1a shows two-inch wafers after the first and second growth steps, respectively, showing good wafer-scale uniformity. The monolayer MoS₂ wafer demonstrates single-crystalline characteristics as evidenced by second-harmonic generation (SHG) mapping and low-energy electron diffraction (LEED; Supplementary Fig. 2). Figure 1c presents an optical microscopy (OM) image of a multilayer 3R-MoS₂ film, in which the first layer is complete and the triangular domains are mostly 2–5 layers. Remarkably, the upper layers exhibit unidirectional alignment regardless of the number of layers. The 3R stacking is confirmed by cross-sectional high-angle annular dark-field (HAADF) STEM imaging of a multilayer region (Fig. 1b and Supplementary Fig. 3), where each layer retains the same orientation but shifts along the (10 $\bar{1}$ 0) direction by $\frac{\sqrt{3}}{3}a = 0.182$ nm. The unit cell contains three layers, and the fourth layer is precisely aligned with the first layer. On the basis of the OM images, we utilize a pixel-level machine learning classification artificial intelligence (AI) model to automatically map out the number of layers and domain orientations (Fig. 1d,e and Supplementary Figs. 4 and 5). For the particular sample shown in Fig. 1c, it comprises 100% monolayer, 98.3% bilayer, 69.8% trilayer, 16.3% quadruple layer, 3.9% pentalayer or more (Fig. 1f). The number of layers obtained by the AI model is further corroborated by Raman spectroscopy, where the separation of the E_{2g} and A_{1g} peaks is 19.2 cm⁻¹ for monolayer and gradually increases to 24.5 cm⁻¹ for the quadruple layer (Fig. 1g). The phase purity of 3R stacking is calculated to be 99.86% from more than 4,000 domains across the entire wafer (the anti-parallel 2H domains are identified by the pink colour in Supplementary Fig. 5). We also perform LEED with a spot size of ~1 mm (Fig. 1h). The LEED pattern shows three diffraction spots, revealing the non-centrosymmetric 3R stacking over a large area.

Spectroscopic and structural characterizations of 3R-MoS₂

Next, we perform systematic spectroscopic and structural investigations of interlayer stacking. Figure 2a shows an OM image of continuous monolayer MoS₂ with unidirectional bilayer domains. The corresponding photoluminescence (PL) image (Fig. 2b) distinguishes

the dark bilayer regions from the bright monolayer regions due to the direct-to-indirect bandgap transition. The atomic force microscopy (AFM) image shows the clean surface of the as-grown MoS₂ (Fig. 2c). SHG offers a direct spectroscopic method for interlayer stacking. Due to the non-centrosymmetric structure of 3R-MoS₂, the SHG intensity shows a quadratic dependence on the number of layers (Fig. 2d,e). By contrast, the 2H stacked bilayer shows negligible SHG signals due to the centrosymmetric structure (Fig. 2d and Supplementary Fig. 6). Raman and SHG mappings further reveal the uniformity of 3R stacking within and across different domains (Supplementary Fig. 7).

Ultralow-frequency (ULF) Raman spectroscopy is another probe to identify the stacking order as the ULF vibration modes arise from interlayer interactions²⁸. *N*-layer MoS₂ possesses *N* – 1 interlayer shear vibrational modes (S modes) and *N* – 1 breathing vibrational modes (B modes)²⁹. The ULF Raman spectra of bilayer 3R-MoS₂ and 2H-MoS₂ are shown in Fig. 2f. In the 3R phase, the peaks at 21.3 cm⁻¹ and 38.7 cm⁻¹ are attributed to the interlayer shear mode (S₁) and breathing mode (B₁), respectively²⁸. In the 2H phase, the S₁ mode shows stronger intensity, and the B₁ mode is blueshifted to 40.7 cm⁻¹, consistent with a stronger interlayer coupling. Furthermore, the redshift in the S_{*N*-1} mode and B_{*N*-1} mode with the number of layers is observed in the 3R configuration (Supplementary Fig. 7c), which is consistent with the theoretical calculations based on the linear chain model²⁸.

The 3R stacking is further revealed at the atomic scale by HAADF-STEM (Fig. 3). Figure 3a,b reveals the atomically sharp monolayer/bilayer and bilayer/trilayer interfaces. Figure 3c–e presents the experimental and simulated HAADF-STEM images of mono-, bi- and trilayer 3R-MoS₂. For monolayer MoS₂, the hexagonal rings, consisting of alternating Mo and S₂ columns, exhibit a clear intensity contrast (Fig. 3c,f). For bilayer MoS₂, the HAADF-STEM image reveals an interlayer sliding by $\frac{\sqrt{3}}{3}a$ along the armchair direction, forming three distinct intensities corresponding to S₂, Mo and Mo + S₂ (Fig. 3d,f). For trilayer MoS₂, further sliding of the third layer results in Mo + S₂ in all the atomic columns, showing very subtle differences depending on which layer the S₂ and Mo atoms reside (Fig. 3e,f). In all cases, simulations based on the 3R-MoS₂ structure quantitatively reproduce the experimental data. We further examine the atomic structure at the domain edges and observe that the transition is atomically sharp (Fig. 3f and Supplementary Fig. 8). These results indicate that the commensurate 3R stacking is already established at the growth front.

Thickness control is another critical aspect of 3R-MoS₂. This can be achieved by the Ostwald ripening process³⁰, in which we periodically interrupt the Mo source to prevent the nucleation of thicker MoS₂ on the top layer. A comprehensive set of characterizations, including OM, AFM and Raman spectroscopy, show large-area uniform bilayer 3R-MoS₂ using this process (Supplementary Fig. 9 and the ‘Discussion about the growth of uniform bilayer MoS₂’ section in the Supplementary Information).

Growth mechanism of 3R-MoS₂

Next, we discuss the growth mechanism of rhombohedral-stacked MoS₂. The selective growth of the 3R-MoS₂ phase necessitates a substantial difference in formation energy between the 2H and 3R phases. However, the energy difference in the bulk form is too small (approximately 1 meV per MoS₂) to account for the selectivity¹⁹. Like on the sapphire substrate, the most critical factor influencing the MoS₂ orientation is nucleation. Once the nuclei reach a critical size, they become energetically unfavourable to rotate freely. Several nucleation mechanisms have been proposed for TMD, namely, impurity nucleation, edge nucleation and surface nucleation. Impurity nucleation (such as Re, W, Nb, V, Fe and Ti) was reported in the chemical vapour transport growth of 3R-MoS₂ bulk crystals^{18,31}. However, it can be ruled out that no metal impurities were introduced during our growth process. Edge nucleation was reported in the epitaxial growth of 2D materials on single-crystal surfaces with atomic steps such as Au and sapphire^{27,32–34}. This mechanism can also

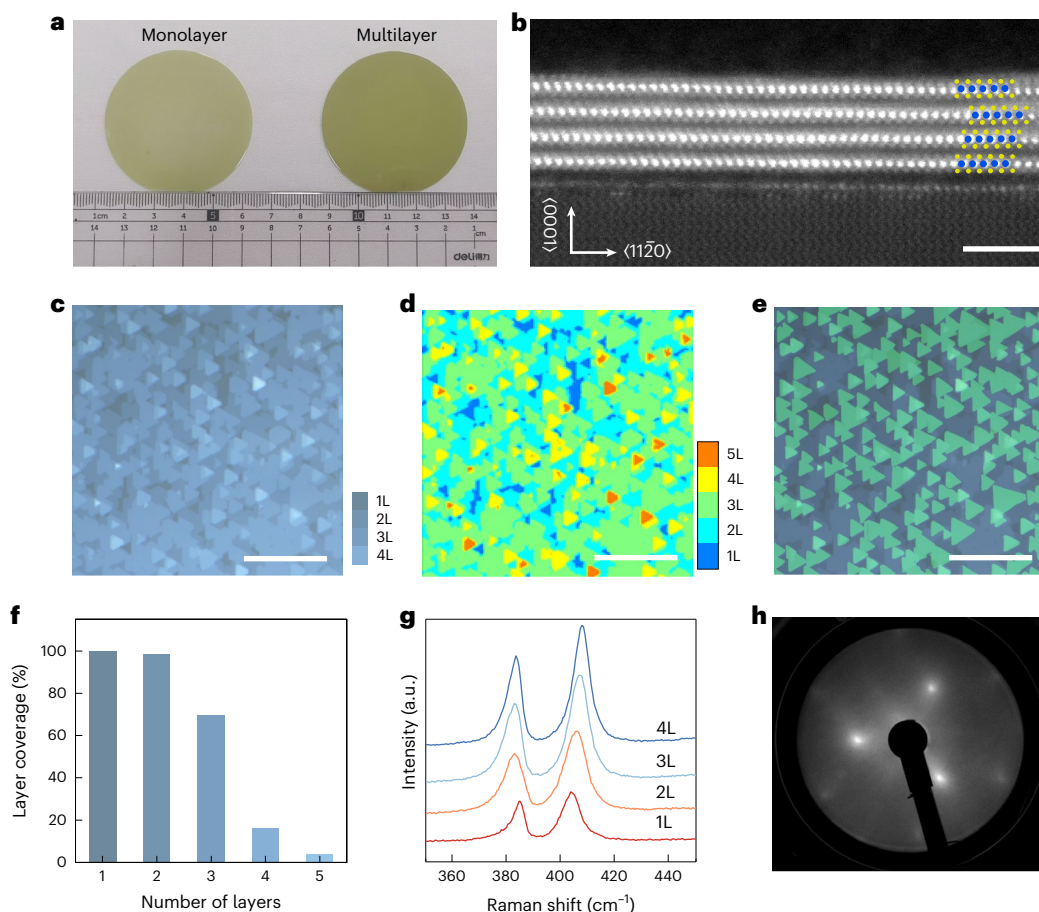


Fig. 1 | Wafer-scale 3R-MoS₂. **a**, Photograph of a monolayer single-crystal MoS₂ film and multilayer 3R-MoS₂ wafers. The optical contrast of the two wafers demonstrates the thickness difference. **b**, Cross-sectional HAADF-STEM image showing a four-layer MoS₂ with 3R stacking. Scale bar, 2 nm. **c**, OM image showing the unidirectional 3R stacking of MoS₂ domains. Scale bar, 40 μm. 1L, one

layer; 2L, two layers; 3L, three layers; 4L, four layers. **d, e**, False-colour diagram of MoS₂ with different layer numbers (**d**) and orientations (**e**) identified by AI corresponding to **c**. Scale bars, 40 μm. **f**, Statistics on the layer coverage via the AI model. **g**, Raman spectra of MoS₂ with different layer numbers transferred on SiO₂/Si. **h**, LEED pattern showing the 3R stacking of the as-grown MoS₂.

be ruled out because the bottom MoS₂ layer was continuous without exposed atomic steps. For surface nucleation, common nucleation centres include grain boundaries and point defects^{35–38}. In our growth process, the first MoS₂ layer is single crystalline without grain boundaries. Therefore, it is reasonable to hypothesize that point defects in MoS₂ act as nucleation sites for 3R-MoS₂.

To obtain experimental evidence of defect-induced nucleation, we carried out a series of control experiments in which the first MoS₂ layer was exposed to ambient atmosphere or oxygen before the growth of the second layer (Methods). We found that after exposure, even using the same growth condition, we could not obtain phase-pure 3R-MoS₂ but instead a mixture of 2H and 3R stackings (Supplementary Fig. 10). The percentage of 3R domains decreased monotonically with the amount of exposure to ~50–60%, where there was little selectivity between 3R and 2H (Supplementary Fig. 11). Although oxygen, water and carbon can be physisorbed on the basal plane of MoS₂ (refs. 39,40, Supplementary Fig. 12 and Supplementary Table 1), the nucleation is not likely to occur at physically absorbed hydrocarbon species as they are removed under the growth conditions (Supplementary Fig. 13 and the ‘Discussion about carbon absorption species on MoS₂’ section in the Supplementary Information). Therefore, these control experiments point to the possible role of lattice defects in modulating the nucleation of the 3R and 2H phases.

To further substantiate our hypothesis, we performed DFT calculations to determine the nucleation energy of bilayer MoS₂ at common defect sites⁴¹, including sulfur vacancies (V_S), double sulfur vacancies

(V_{S2}), Mo₅ and Mo vacancies (V_{Mo}) (Fig. 4a). These are the main defect types observed in our monolayer and bilayer MoS₂ (Supplementary Fig. 14). For each structure, we calculated the nucleation energy for 3R and 2H stackings and the energy difference ($E_{F-3R} - E_{F-2H}$) was used as a measure of selectivity. As shown in Fig. 4b, most defects have near-zero $E_{F-3R} - E_{F-2H}$, indicating no selectivity between 3R and 2H. The only defect favouring the nucleation of 3R over 2H is Mo₅ anti-site with $E_{F-3R} - E_{F-2H} = -75$ meV per MoS₂. We further considered the passivation of Mo₅ in ambient oxygen by calculating various passivated structures (Mo₅-O, Mo₅-O₂, Mo₅-OH, Mo₅-H and Mo₅-H₂O; Fig. 4a). The result shows that $E_{F-3R} - E_{F-2H}$ is significantly reduced by passivation, which explains the reduced 3R ratio in control experiments (Supplementary Figs. 10 and 11).

Further evidence of defect-promoted nucleation was obtained by the HAADF-STEM measurement of bilayer domains at the initial nucleation stage (Fig. 4c, d). Figure 4c, d shows the STEM image of a typical 3R bilayer domain with a diameter of ~7 nm, where a single, bright atomic defect is observed. Through experimental and simulated line profile analysis, we identify the defect site as Mo + Mo₅ + S (Fig. 4e), suggesting that the defect is Mo₅ in the first layer. For larger domains, we also observe Mo substitution sites near the centre (Supplementary Figs. 15 and 16). Therefore, we speculate that nucleation is most likely related to this defect type (more discussions are provided in the ‘Discussion about the nucleation mechanism of 3R-MoS₂’ section in the Supplementary Information). Moreover, Mo₅ defects were observed at the centre of trilayer 3R domains (Supplementary Fig. 17). By contrast, no Mo₅

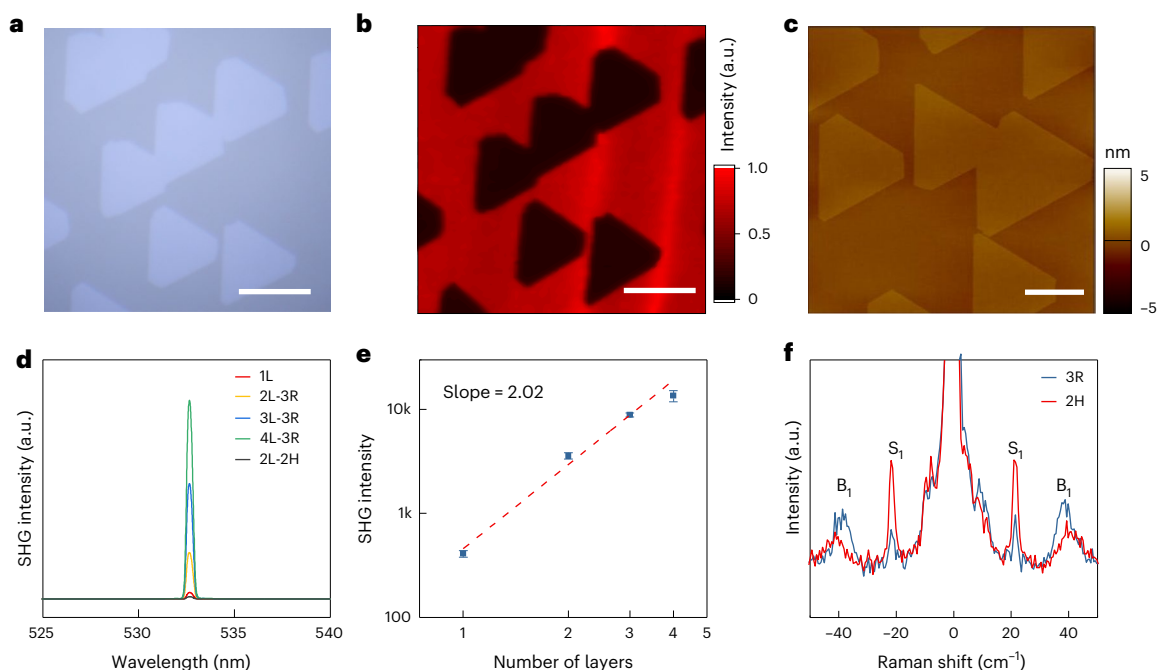


Fig. 2 | Spectroscopic characterizations of 3R-MoS₂. **a**, OM image of 3R-stacked bilayer MoS₂, which is composed of continuous monolayer and unidirectional, discrete bilayer domains. **b**, Corresponding PL mapping to **a**, showing the direct band emission of the monolayer zone and indirect band of the epitaxial bilayer domains. Scale bars, 10 μ m (**a** and **b**). **c**, AFM height image demonstrating the

high surface quality. Scale bar, 6 μ m. **d**, Layer-number-dependent SHG spectra. **e**, Linear fitting of the SHG intensity with the layer number. Data are presented as mean values \pm standard deviation. The statistical data are from ten samples. **f**, ULF Raman spectra of bilayer MoS₂ with 2H and 3R stacking.

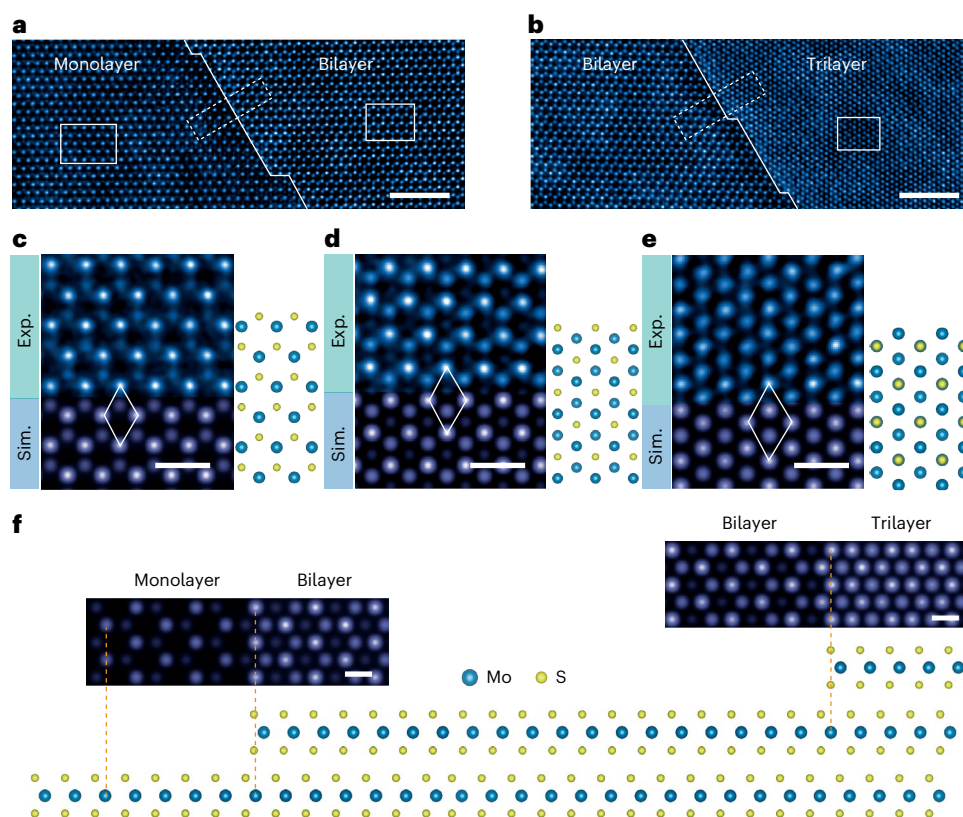


Fig. 3 | Atomic-resolved characterization of 3R-stacked MoS₂. **a, b**, Domain edge regions between mono-/bilayer R stackings (**a**) and bi-/trilayer stackings (**b**) with sharp edges. Scale bar, 2 nm. **c–e**, Experimental (Exp.) high-resolution HAADF-STEM images (left) and simulated (Sim.) images (right) of monolayer-

stacked MoS₂ (**c**), bilayer-stacked MoS₂ (**d**) and trilayer-stacked MoS₂ (**e**). Experimental areas correspond to the regions marked with the white boxes in **a** and **b**. Scale bars, 0.5 nm. **f**, Simulated images of the interface between different layers and a schematic of the side view of the 3R configuration. Scale bars, 0.2 nm.

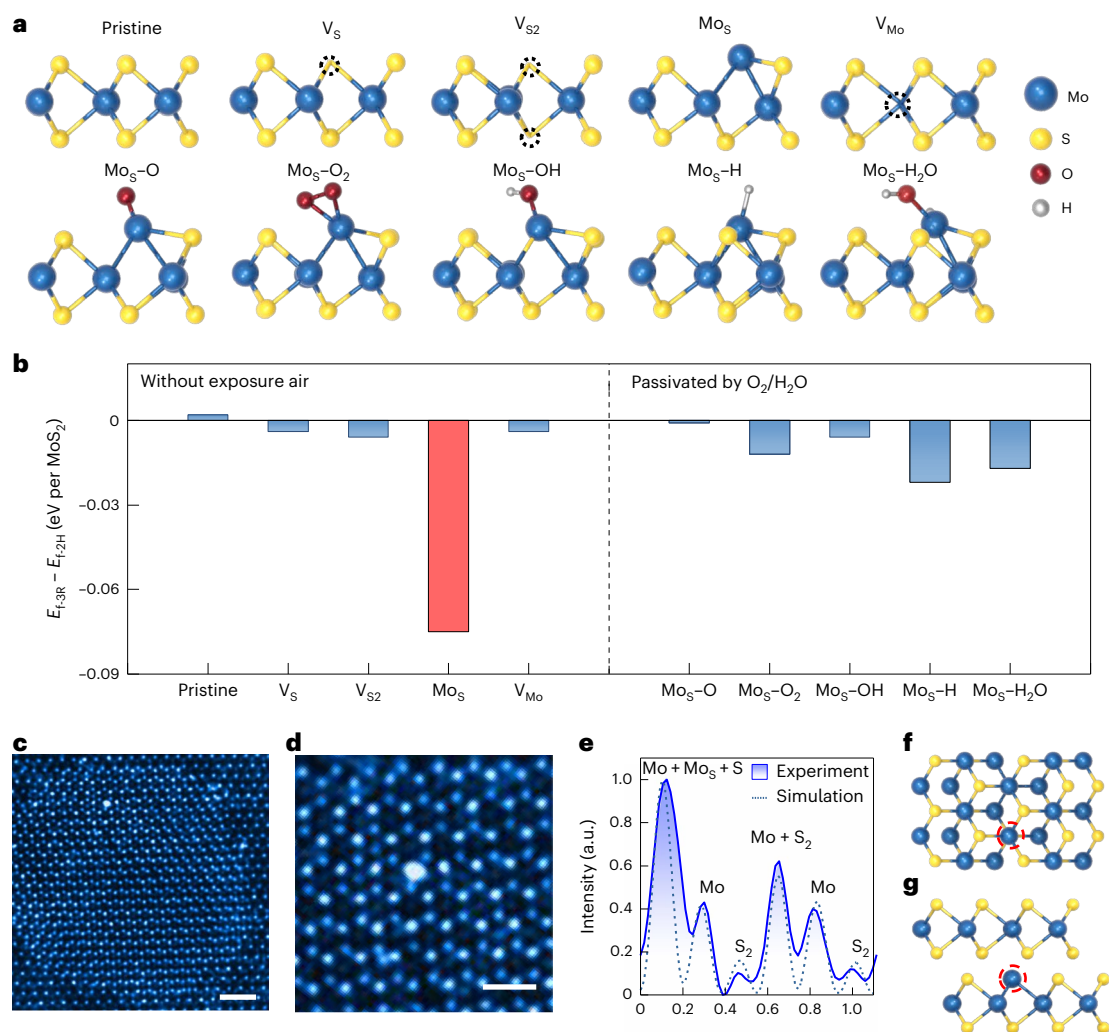


Fig. 4 | Mechanism of the selective growth of 3R-phase MoS₂. **a**, Atomic structures of pristine MoS₂ and defects, V_S, V_{S2}, Mo_S, V_{Mo} and Mo_S defects with absorption from oxygen and water when monolayer MoS₂ is exposed to ambient atmosphere. **b**, Formation energy difference between 3R and 2H stacking orders of bilayer MoS₂ with different defect types being presented on monolayer MoS₂. **c**, HAADF-STEM images of bilayer nucleation. **d**, A magnified image in **c**, which

shows the MoS nucleation centre. The white dashed box in **c** corresponds to the regions where **d** resides. There is a brighter atomic site in the centre. **e**, Line profile analysis of the defect site along with the simulation results. **f, g**, Schematic of the top (**f**) and side (**g**) views of Mo_S defect-induced nucleation. Scale bars, 1 nm (**c**); 0.5 nm (**d** and **e**).

defects were found in the 2H domains (Supplementary Fig. 18). Other defect types beyond our calculations cannot be completely ruled out at this stage, which will be explored in future works.

Ferroelectricity and device applications of 3R-MoS₂

The broken inversion symmetry in 3R-MoS₂ gives rise to ferroelectricity. Prior works have demonstrated sliding ferroelectricity in mechanically exfoliated 3R-MoS₂ and twisted moiré structures^{5–7}, but CVD films have been rarely explored for device applications²⁰. We first perform piezoresponse force microscopy (PFM) to verify the ferroelectricity in bilayer 3R-MoS₂ films at room temperature (Fig. 5a and Methods). Figure 5b,c shows the AFM height and the corresponding PFM phase images of a bilayer domain. Although the AFM shows uniform height across the bilayer region, the PFM phase shows two distinct colours, indicating ferroelectric domains with different polarization directions (more ferroelectric data are shown in Supplementary Fig. 19). The ferroelectric domains were revealed under tilted dark-field transmission electron microscopy (TEM; Supplementary Fig. 20)⁸. The sharp boundaries between the domains correspond to domain walls. The ferroelectric

polarization switching is measured by an off-field piezoresponse loop in PFM (Fig. 5d). The amplitude and phase show typical butterfly and hysteresis loop features, respectively, with ± 5 -V coercive voltage and 180° phase shift. Different sample locations show very good reproducibility of ferroelectric switching (Supplementary Fig. 21). The out-of-plane piezoelectric coefficient d_{33} is measured to be -0.31 pm V^{-1} for bilayer 3R-MoS₂ (Supplementary Fig. 22), in agreement with mechanically exfoliated samples⁴².

As a ferroelectric semiconductor channel, 3R-MoS₂ can be used to build FeS-FETs with a non-volatile memory functionality. To demonstrate the unique advantage of large-area 3R-MoS₂ films, we fabricated device arrays using back-gate structures (Fig. 5e,f). Figure 5g plots the transfer characteristics of 48 FeS-FETs. The devices show a counter-clockwise hysteresis loop, a signature of ferroelectric memory. Due to the uniform 3R nature of the film, the devices exhibit good reproducibility with a narrow distribution of memory window ($2.34 \pm 0.46 \text{ V}$; Fig. 5h). By contrast, monolayer MoS₂ FETs fabricated using the same process show much more negligible clockwise hysteresis (Supplementary Fig. 23a), which is attributed to interface traps rather than ferroelectric switching.

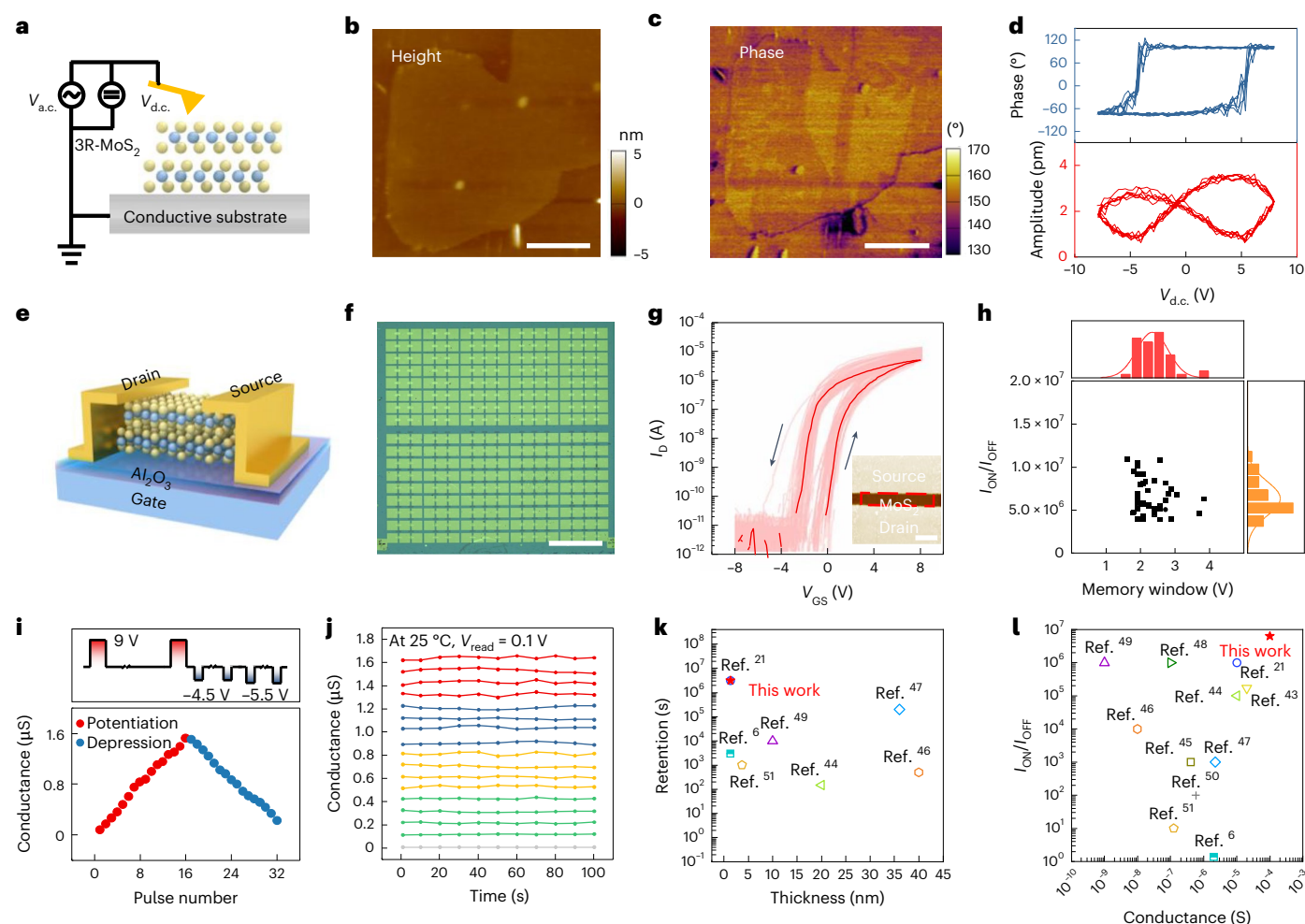


Fig. 5 | Ferroelectricity of R-stacked bilayer MoS₂. **a**, Schematic of vertical PFM measurements. **b**, AFM image of a bilayer flake on a monolayer film. Scale bar, 2 μm . **c**, Phase image of vertical PFM under $V_{\text{ac}} = 2\text{ V}$ corresponding to **b**. Scale bar, 2 μm . **d**, Amplitude and phase of the piezoresponse signal versus sample bias in vertical PFM. **e**, Schematic of a 3R-MoS₂ FeS-FET. **f**, OM image of an FeS-FET array. Scale bar, 300 μm . **g**, Transfer characteristics of the 3R-MoS₂ FeS-FET at sweep ranges of $V_{\text{GS}} = \pm 8\text{ V}$, showing an obvious counterclockwise hysteresis behaviour. The inset shows the AFM image of FeS-FET. Scale bar, 3 μm .

h, Statistical distribution of the memory window and $I_{\text{ON}}/I_{\text{OFF}}$. **i**, Symmetric and linear multistate properties of the 3R-MoS₂ FeS-FET; the operation waveform are shown at the top of the figure. **j**, Multistate retention of 3R bilayer MoS₂ FeS-FET. **k**, Benchmark of the retention of FeS-FETs with different channel thicknesses. The FeS-FET in this work possesses a channel with the thinnest thickness but the highest retention. **l**, Benchmark of $I_{\text{ON}}/I_{\text{OFF}}$ versus the conductance of FeS-FETs in the literature.

We further measure the retention characteristics of the FeS-FET (Supplementary Fig. 23b). The device is programmed to the ON/OFF state by $\pm 10\text{-V}$ voltage pulses, followed by a d.c. sampling test for thousands of seconds. The decay of the resistive states is minimal within the test duration. By extrapolating the retention data, the ON/OFF ratio still exceeds three orders of magnitude after 10 years (Supplementary Fig. 23b). Moreover, the FeS-FET devices show good multibit characteristics. Figure 5i,j shows the potentiation and depression process of 16 storage states (4 bit) with good symmetry, linearity and retention, demonstrating the potential application of 3R-MoS₂ in analogue neuromorphic computing.

Finally, we benchmark our device performance with other 2D materials from the literature, including In₂Se₃, InSe, SnS and sliding 2D ferroelectric materials such as 3R-TMDs and twisted TMDs (Fig. 5k,l and Supplementary Table 2)^{6,21,43–51}. Figure 5k plots the retention versus thickness. Our data are on the preferred top left corner with the best retention despite the smallest thickness of 1.3 nm. We should emphasize that all the other devices in the top left corner are based on exfoliated flakes rather than large-area CVD films. Figure 5l compares the ON-state conductance and ON/OFF ratio. Our device lies in

the preferred top right corner with both high ON-state conductance and ON/OFF ratio. Overall, our FeS-FET devices exhibit one of the best ferroelectric device performances among 2D materials and offer a large-area device array. This makes the CVD 3R-MoS₂ very promising for device applications. We note that not all the FeS-FET devices exhibit ferroelectric memory behaviour, and the related discussions are provided in the ‘Discussion about FeS-FET of 3R-MoS₂’ section in the Supplementary Information.

Conclusion

In conclusion, we demonstrate the CVD growth of rhombohedral-stacked large-area MoS₂ films via homoepitaxy. A defect-promoted selective nucleation mechanism is proposed, which may inspire future research on growing artificially stacked 2D materials. The control of interlayer stacking by CVD offers a degree of freedom in the production of large-area 2D materials towards multifunctional integration.

Online content

Any methods, additional references, Nature Portfolio reporting summaries, source data, extended data, supplementary information,

acknowledgements, peer review information; details of author contributions and competing interests; and statements of data and code availability are available at <https://doi.org/10.1038/s41563-025-02274-y>.

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Methods

CVD growth of 3R-MoS₂

Single-side polished sapphire (0001) with a miscut angle of 1° towards the *a* axis (*c/a* = 1°) was used as the epitaxial substrate. Before growth, the substrates were subject to an annealing process at 1,000–1,100 °C for 4 h in air, which produces uniform bi-steps, as previously reported^{19,27}. The growth was performed in a three-zone CVD system. Elemental sulfur (30 g, 99.995%) was utilized as the source and placed in a silica crucible and heated independently with an extra heating mantle at 180 °C and carried by 150 s.c.c.m. of Ar. Metallic Mo foils were used as the Mo sources and put in a separate tube from S. This structural design can avoid the contamination of Mo sources by S vapours. Most importantly, the Mo foils were oxidized in situ by 5–10 s.c.c.m. of O₂ to release MoO₃ vapours and carried by 50 s.c.c.m. of Ar into the growth chamber, which is a controllable way and ensures a stable, precise and continuous supply of sources. Since Mo metal has a very high melting temperature (2,620 °C), unoxidized Mo cannot evaporate at the heating temperature. When we switch on the O₂ flux, Mo can be oxidized into MoO₃, which can evaporate into the CVD system.

The growth comprises two steps. In the first step, a single-crystal monolayer film is epitaxially produced on the sapphire substrate at 950 °C. In the second step, the concentration of the Mo species was improved to boost the nucleation of the top layers by elevating the temperature of the Mo source from 600 °C to 800 °C and keep the growth temperature constant. According to our measurements, the MoO₃ vapour flux was increased from 5.8 μg min⁻¹ (600 °C) to 110 μg min⁻¹ (800 °C). The growth pressure is ~90 Pa and the whole growth process lasts about 30–60 min. The purity of the carrier gas sources is more than 99.999%. We note that O₂ mostly reacted with S vapour to form SO₂ during the growth, as measured by in situ mass spectrometry (Supplementary Fig. 25). The measured O₂ partial pressure of 0.495 Pa corresponded to chemical potential Δμ_O = -2 eV. The formation energy of O₂ passivating Mo_s defects at Δμ_O = -2 eV is much higher (by 3.42 eV) than in the air (Supplementary Fig. 26), indicating that Mo_s cannot be passivated by O₂ during growth.

The transfer of MoS₂ was done using the polymer-based transfer method. Poly(methyl methacrylate) (PMMA; Allresist 950K) was spin coated on MoS₂/sapphire at 2,000 rpm and baked at 150 °C. Then, a thermal release tape (TRT) was laminated and used to gently peel the PMMA/MoS₂ stack off the sapphire substrate in KOH water. The TRT/PMMA/MoS₂ stack was transferred onto the target substrate or a TEM grid and baked at 120 °C to release the TRT. Finally, the PMMA was removed by acetone.

Control experiments with the first MoS₂ layer exposed in ambient atmosphere or oxygen

After the monolayer MoS₂ growth was completed and cooled to room temperature, we kept it in the ambient atmosphere for 5 min, 3 h and 12 h. Then, the monolayer film was used as a template for the second-layer growth. For oxygen exposure, we introduced a mix of O₂ and Ar into the growth chamber after the single-crystal film cooled to room temperature. This mixture was maintained for 1 h with different O₂/Ar ratios to allow for O₂ adsorption under a total pressure of 1 bar. Afterwards, we pumped the chamber into a vacuum and proceeded with the next growth process.

Computation details

All the first-principles DFT calculations utilized the projector-augmented wave⁵² potentials implemented in the Vienna ab initio simulation package⁵³. The Perdew–Burke–Ernzerhof⁵⁴ generalized gradient approximation was used to treat the exchange–correlation functional. Grimme's DFT-D3 (ref. 55) scheme was used to characterize the van der Waals interaction, and the Brillouin zone was solely sampled at the gamma point. The structures underwent relaxation until the residual Hellman–Feynman force was <0.02 eV Å⁻¹, with an energy

convergence criterion of 10⁻⁴ eV. A 400-eV energy cut-off was selected for the plane-wave basis sets. Additionally, to prevent interaction between adjacent images, a vacuum space of >10-Å thickness was included in the vertical direction. A 1 × 2 × 1 *k*-point mesh was used for MoS₂ triangular clusters stacking on the first layer of MoS₂. The formation energy of 3R and 2H configurations for monolayer MoS₂ with a specific defect is calculated by $E_f = \frac{E_{\text{total}} - E_{\text{2L-cluster}} - E_{\text{1L-MoS}_2}^{\text{defect}}}{N_{\text{2L-MoS}_2}}$, where E_{total} , $E_{\text{2L-cluster}}$ and $E_{\text{1L-MoS}_2}^{\text{defect}}$ are the calculated energies of bilayer MoS₂ clusters stacking on monolayer MoS₂ with a specific defect, bilayer MoS₂ clusters and monolayer MoS₂ with a specific defect, respectively; and $N_{\text{2L-MoS}_2}$ is the number of MoS₂ units of bilayer MoS₂ clusters. In our calculations, the length of bilayer MoS₂ clusters is ~0.95 nm and $N_{\text{2L-MoS}_2} = 12$. Each supercell contains a single defect and different defect sites are considered when stacking the second layer.

The formation energies of Mo_s and Mo_s-O₂ defects are defined by

$$E_f = E_{\text{defect}} - E_{\text{MoS}_2} + \mu_S - \mu_{\text{Mo/O}}, \quad (1)$$

where E_{defect} and E_{MoS_2} are the energies of MoS₂ with and without defects, respectively; μ_S and $\mu_{\text{Mo/O}}$ are the chemical potentials of S and added Mo/O, respectively.

The chemical potential of oxygen at the given partial pressure (*P*) and temperature (*T*) can be expressed as below.

$$\begin{aligned} \Delta\mu_{\text{O}(\text{O}_2)}(T, P) &= \mu_{\text{O}(\text{O}_2)}(T, P) - \mu_{\text{O}(\text{O}_2)}(0, P^0) \\ &= \frac{1}{2} \left(H_{\text{O}_2}^0(T) - TS_{\text{O}_2}^0(T) - H_{\text{O}_2}^0(0) \right) + \frac{1}{2} RT \ln \left[\frac{P_{\text{O}_2}}{P^0} \right] \end{aligned} \quad (2)$$

According to the partial pressure of O₂ in the furnace during growth, the chemical potential of O is -2 eV, whereas Δμ_O = -0.296 eV in the air at room temperature.

Optical characterizations

PL and Raman spectroscopy and mapping were performed by a custom-built system equipped with a 488-nm laser excitation and Princeton Instruments SP-2500 spectrometer. SHG spectroscopy and mapping were carried out in a customized system equipped with a piezo stage, an ultrafast 1,064-nm-wavelength laser (Rainbow 1064 OEM) and a photon-counting head (Hamamatsu Photonics H7421-50). A confocal micro-Raman spectrometer (Horiba LabRAM HR Evolution) with an 1,800-lines-per-mm grating was used to test the ULF Raman data.

PFM measurements

All the PFM measurements were adapted to the Dart-SS-PFM mode. The vertical PFM mapping, hysteresis loop and KPFM measurements were performed on the bilayer 3R-MoS₂ that was transferred onto the Pt substrate by PMMA. PFM measurements were conducted with Asylum Research Cypher SAFM device at room temperature. We used Asytec-01-R2 with a force constant of around 2.8 N m⁻¹ and a contact resonance frequency of around 280 kHz with the applied a.c. bias voltage of 2 V for the Dart-SS-PFM measurements.

Vertical PFM (VPFM) captures the vertical signal of the cantilever. In real VPFM measurements, the recorded VPFM signal is fairly complex, arising from the interplay among piezoelectricity, electrostatic interaction⁵⁶, electrochemical strain⁵⁷ and even the experimental setup⁵⁸. All those mechanisms may contribute as a background VPFM signal to distort a relatively small VPFM response of the sample. To obtain the pure electromechanical response of 3R-MoS₂, we use vector decomposition to remove the background signal. A mask can be manually sketched based on phase and amplitude maps to distinguish the 3R-MoS₂ area and monolayer MoS₂ (background) area. The background VPFM signal of one scan line (Åe¹⁰) is calculated by the mean measured VPFM vector of monolayer MoS₂ area ($A_m e^{i\theta_m}$):

$$\bar{A}e^{i\theta} = \frac{1}{n} \sum_{j=1}^n A_{m,j} e^{i\theta_{m,j}}; \quad (3)$$

the decoupled VPFM vector for every pixel ($A_d e^{i\theta_d}$) is determined by subtracting this background VPFM vector from the measured VPFM vector ($A_m e^{i\theta_m}$):

$$A_d e^{i\theta_d} = A_m e^{i\theta_m} - \bar{A} e^{i\theta}, \quad (4)$$

and the corresponding decoupled phase (θ_d) and amplitude (A_d) maps are shown in Supplementary Fig. 22.

TEM characterization

Epitaxial 3R-MoS₂ was transferred onto a TEM grid (Cu grid with a carbon array microgrid). The atomic-resolved HAADF-STEM and cross-sectional HAADF-STEM of different layers were captured on FEI Themis Z at an accelerating voltage of 300 kV. The in-plane HAADF-STEM images were filtered by Gaussian filters. The tilt dark-field TEM images were captured on an aberration-corrected STEM Titan Cubed G2 60-300 system with an accelerating voltage of 60 kV. All the in-plane STEM simulations used scripts based on the modified MULTTEM package using MATLAB R2020b and accelerated by GPU calculations⁵⁹. The STEM image simulations are run at 300 kV with an aberration-free probe and a convergence angle of 22.5 mrad. The scanning pixel size is 0.1 Å. STEM detector angle is set to 75–200 mrad, which is consistent with the experiment condition. The source size and nonlinear effect of the detector are considered in the simulation.

Device fabrication and measurement

Here 20-nm Al₂O₃ was deposited by plasma-enhanced atomic layer deposition as a dielectric insulator at 150 °C using O₂ plasma as the oxygen source and trimethyl aluminium as the precursors on a heavily doped silicon substrate as the global gate. After that, bilayer MoS₂ was transferred onto it by using TRT. Then, electron-beam lithography and reactive ion etching with O₂ plasma were used to pattern the MoS₂ nanoribbons. The contact electrodes were defined by electron-beam lithography, and 15 nm of Sb and 30 nm of Au were finally deposited as the source and drain, to suppress Fermi-level pinning and reduce the contact resistance⁶⁰. Electrical measurements were performed at room temperature using a Keithley 4200 semiconductor characterization system with four source meter units for the d.c. test and 4,225 remote pulse and switch modules for the pulse test under a base pressure of 10^{−6} torr in the Lakeshore CRX-VF probe station. With source meter units, the transfer and output properties were characterized. The MoS₂ FET was programmed to the ON state (or erased to the OFF state) with 10 V/2 s (or −10 V/2 s); then, a d.c. sampling test proceeded for thousands of seconds, thereby extrapolated to 10 years of retention by linear fitting. As for the multistate test, the potentiation (or depression) process was realized by identical 9 V/40 ms (or −4.5 V/5 ms for the first eight states and −5.5 V/5 ms for the second eight states) generated by 4,225 remote pulse and switch modules at the gate, whereas the drain and source of the FET were both grounded. Different pulse conditions in depression progress were applied for linear weight update⁶¹. Afterwards, the typical output curves were obtained under a small V_{DS} (±0.2 V) range to read the conductance of MoS₂ FET.

AI recognition of MoS₂ layer numbers

Recognizing the limitations posed by the background tilt, which affects the direct use of brightness pixel counts for layer determination, we use a machine-learning-based approach enhanced by a comprehensive calibration process for determining the layer numbers of 2D materials viewed using OM. This process adjusts for variations in lighting conditions, ensuring accurate feature extraction critical for layer identification. Our calibration procedure extends beyond the traditional focus on image corners to include the edges and the centre, capturing

a comprehensive range of brightness levels across the image. This thorough calibration ensures that the adjustments made for lighting conditions are uniformly applied, thereby rectifying the illumination inconsistencies inherent in the OM of 2D materials. Following calibration, the process proceeds with the extraction of features critical to our analysis, including the pixel coordinates (X , Y) and RGB values. These extracted features serve as inputs to our machine learning model, facilitating accurate predictions of layer numbers across the sample.

In the selection of an appropriate machine learning model, we leverage an automated machine learning function to evaluate various algorithms, ultimately selecting a support vector machine model for its superior performance and captivity at handling nonlinear relationships and complex patterns present in our MoS₂ OM images (Supplementary Fig. 4a). This model is trained on a labelled dataset, where each pixel is associated with a known layer number, ensuring a robust framework capable of generalizing from our training data to accurately predict layer numbers in new, unseen images. Utilizing the trained support vector machine model, we predict the layer number for each pixel in the image, allowing for a detailed mapping of layer distribution across the sample. This approach enables the precise identification of different material layers, offering a granular view of the sample's composition.

A detailed map displaying each layer's distribution, colour coded for clarity and ease of interpretation. The model's confusion matrix highlights the accuracy and precision in predicting various layer numbers, thereby providing an insight into the model's effectiveness (Supplementary Fig. 4b). Distribution histograms quantify the percentage coverage of each layer within the samples, alongside cumulative histograms that offer a nuanced understanding of overall layer distribution.

Our analytical approach not only underscores the efficacy of combining image calibration, feature extraction and machine learning in accurately determining the layer numbers but also facilitates a deeper understanding of the 2D materials examined. By leveraging comprehensive calibration alongside a sophisticated support vector machine model selected through automated machine learning, our methodology represents a significant advancement over traditional analysis techniques, offering enhanced accuracy and insight into the material properties.

MoS₂-domain orientation analysis and image segmentation model

To accurately and efficiently distinguish the orientation of MoS₂ crystals and apply pseudocolor processing to crystals with different orientations, an image segmentation model based on deep learning is considered. The principle of image segmentation involves dividing the image into several subregions that share similar colour or texture characteristics, which correspond to different objects or different parts of objects. These subregions, constituting the complete subset of the image, are mutually exclusive. Traditional segmentation methods are represented by graph-cut, whereas deep learning models include U-Net, FCN and DeepLab⁶². In this paper, the U-Net model for image segmentation is applied, and the principle of this model and its segmentation results are introduced below.

The U-Net network is composed of two parts: the first half is dedicated to feature extraction (with MobileNet as the base model), and the second half involves upsampling. This structure is also referred to as an encoder–decoder architecture in some studies. The overall structure of this network resembles the uppercase letter 'U', which is the reason behind its name, U-Net. A diagram of its structure is presented in Supplementary Fig. 4c.

The overall structure of U-Net can be divided into two parts: an encoder and a decoder (Supplementary Fig. 4d). The encoder is responsible for progressively extracting features from the input image, reducing the image resolution, whereas the decoder gradually restores the resolution and generates the final segmentation result.

The encoder part of U-Net consists of multiple downsampling layers, each comprising a convolutional layer followed by a pooling layer, which is used to progressively reduce the image size and extract features. This approach is aimed at incorporating context information at different scales. The downsampling section can be considered a conventional convolutional neural network where features are gradually reduced, and spatial information is discarded. This section is composed of two convolution operations and two max-pooling operations, with a rectified linear unit activation function following each convolutional operation. After each max-pooling operation, the width and height of the feature map are halved.

The decoder part of U-Net includes multiple upsampling layers and corresponding skip connections. Upsampling layers are used to gradually increase the image size, whereas skip connections are used to link the feature maps from the encoder to the corresponding feature maps in the decoder, preserving more spatial details and semantic information. Each upsampling layer also has a corresponding convolutional layer for further feature processing. The upsampling section gradually restores the features to their original size and reintegrates lost spatial information back into the features. This section consists of an upsampling operation and a convolution operation, with each convolution operation followed by a rectified linear unit activation function, too.

Within these two parts, numerous skip connections link features from the downsampling section to the upsampling section, allowing for the segmentation to utilize both coarse and fine features. These skip connections enable U-Net to better preserve spatial information, thereby enhancing the accuracy of segmentation.

The loss function of U-Net: U-Net is utilized for image segmentation, where our objective is to assign each pixel in the input image to its respective category. We can use cross-entropy as the loss function, which is defined as follows⁶³:

$$\text{Loss} = -\frac{1}{N} \sum_{i=1}^N \sum_{j=1}^C y_{ij} \log[p_{ij}],$$

where N denotes the number of samples, C represents the number of categories, y_{ij} is the true label indicating whether the i th sample belongs to the j th category, and p_{ij} is the probability output by the network that the i th sample belongs to the j th category. Furthermore, during training, it is necessary to use the backpropagation algorithm to calculate gradients and update the network parameters.

For an AFM image, the encoder of the convolutional neural network, U-Net, is first used to extract the features, obtaining high-level semantic feature maps; then, the decoder restores the feature map to its original size. During the training phase, a loss function is constructed using the model's predicted map and the sample's true label map, thereby facilitating model training. In the inference phase, the model's predicted map serves as the final output. A distinctive feature of U-Net, setting it apart from other common segmentation networks, is its unique feature fusion method—concatenation. U-Net uses concatenation to combine features along the channel dimension, forming denser features⁶⁴.

In U-Net, the pooling layer enables multiscale feature recognition of the image by the network. The upsampling section integrates outputs from the feature extraction segment, effectively fusing multiscale features. For example, the features of the last upsampling originate both from the output of the first convolutional block (same-scale features) and from the output of the upsampling (larger-scale features). Such connections, which occur throughout the network, result in up to four instances of fusion. The modified segment model is able to judge the MoS₂ crystal orientation within 0.1 s, reaching an accuracy of >98%.

Data availability

All data are available in the article and its Supplementary Information. Source data are provided with this paper.

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Author contributions

X.W. conceived and supervised the project. L.L. performed the CVD growth under the supervision of T.L., with assistance from S.Z. and X.Z. X.G., L.M. and J.W. performed the DFT calculations. M.F., H.Z. and

N.Z. performed the AI identification of layer thickness and 3R ratios. L.Z., X.H., S.W. and K.Z. performed the TEM characterization and data analysis. Z.D. performed the STEM simulation. L.L., F.Z. and Z.H. performed the PL, Raman and SHG tests and sample transfer. S.F. and W.X. performed the ULF Raman test. Y.L., G.L., L.T. and Y.J. contributed to the ferroelectric characterization and data analysis. H.W. and Z.Y. fabricated and tested the MoS₂ FeS-FETs. L.L., T.L., L.M., J.W. and X.W. co-wrote the paper with input from the other authors. All authors contributed to the discussions.

Competing interests

The authors declare no competing interests.

Additional information

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